

Novel Activity of *Escherichia coli* Mismatch Uracil-DNA Glycosylase (Mug) Excising 8-(Hydroxymethyl)-3,*N*⁴-ethenocytosine, a Potential Product Resulting from Glycidaldehyde Reaction[†]

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Received July 23, 2001; Revised Manuscript Received December 11, 2001

ABSTRACT: Glycidaldehyde is an industrial chemical which has been shown to be genotoxic in in vitro experiments and carcinogenic in rodent studies. It is a bifunctional alkylating agent capable of reacting with DNA to form exocyclic hydroxymethyl-substituted ethenobases. In this work, 8-(hydroxymethyl)-3,*N*⁴-etheno-2'-deoxycytidine (8-HM- ϵ dC), a potential nucleoside derivative of glycidaldehyde, was synthesized using phosphoramidite chemistry and site-specifically incorporated into a defined 25-mer oligodeoxynucleotide. The 8-HM- ϵ C adduct is structurally related to 3,*N*⁴-ethenocytosine (ϵ C), a product of reaction with vinyl chloride or through lipid peroxidation. In *Escherichia coli*, ϵ C has been shown previously to be a primary substrate for the mismatch uracil-DNA glycosylase (Mug). In this study, we report that the same glycosylase also acts on 8-HM- ϵ C in an oligonucleotide duplex. The enzyme binds to the 8-HM- ϵ C-oligonucleotide to a similar extent as the ϵ C-oligonucleotide. The Mug excision activity toward 8-HM- ϵ C is \sim 2.5-fold lower than that toward the ϵ C substrate. Both activities can be stimulated up to \sim 2-fold higher by the addition of *E. coli* endonuclease IV. These two adducts, when mispaired with normal bases, were all excised from DNA by Mug with similar efficiencies. Structural studies using molecular simulations showed similar adjustment and hydrogen bonding pattern for both 8-HM- ϵ C•G and ϵ C•G pairs in oligomer duplexes. We believe that these findings may have biological and structural implications in defining the role of 8-HM- ϵ C in glycosylase recognition/repair.

Simple epoxides represent an important group of industrial chemicals. Some of these compounds are mutagenic or carcinogenic. For example, glycidaldehyde, a highly reactive epoxide (Figure 1), has been shown to be both mutagenic in in vitro genotoxicity tests (1) and carcinogenic in long-term rodent skin cancer studies (2, 3). On the basis of these studies and others, the International Agency for Research on Cancer (IARC) classified glycidaldehyde as an animal carcinogen (4, 5). Studies from National Toxicology Program (NTP) also showed sufficient evidence for the carcinogenicity of glycidol in experimental animals and anticipated that these biological effects could occur in humans exposed to the compound (6). Glycidaldehyde can be produced from oxidative metabolism of glycidyl ethers (7, 8). Several glycidyl ethers have also been shown to be carcinogenic in experimental animals (9, 10).

Because of its reactive carbonyl and epoxy functionalities, glycidaldehyde is capable of forming cyclic hydroxymethyl-substituted etheno adducts. The structures of the dA and dG adducts formed after reaction with glycidaldehyde have previously been identified and well-characterized (11–16).

[†] This work was supported by NIH Grants CA72079 (to B.H.) and CA47723 (to B.S.) and was administered by the Lawrence Berkeley National Laboratory under Department of Energy Contract No. DE-AC03-76SF00098.

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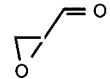
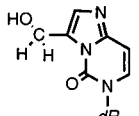
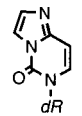
Carcinogen	Metabolite	Adduct Structure
Glycidyl ethers	 Glycidaldehyde	 8-(hydroxymethyl)- 3, <i>N</i> ⁴ -etheno-dC
Vinyl chloride	$\text{ClCH}_2\text{-CHO}$ Chloroacetaldehyde	 3, <i>N</i> ⁴ -etheno-dC

FIGURE 1: Chemical structures of 8-(hydroxymethyl)-3,*N*⁴-etheno-deoxycytidine (8-HM- ϵ dC) and 3,*N*⁴-ethenodeoxycytidine (ϵ dC).

The modified dA nucleoside was identified by Steiner and colleagues in the skin of C3H mice treated with glycidaldehyde (13) or bisphenol A diglycidyl ether (8). The only report on the chemical formation of dC-glycidaldehyde adducts is from Kohwi (17), who observed that glycidaldehyde is highly reactive with dC in non-B DNA in vitro. However, no structural information was given on the dC modification.

To understand the role of a dC adduct in the mechanism of mutagenicity/carcinogenicity of glycidaldehyde as well

as its chemical/physical properties, we recently synthesized 8-(hydroxymethyl)-3, N^4 -etheno-2'-deoxycytidine (8-HM- ϵ dC) (Figure 1) and its phosphoramidite (18). The latter compound was then site-specifically incorporated into defined oligodeoxynucleotides. Using an in vitro replication assay, we have found that the presence of 8-HM- ϵ C¹ either causes blocking of replication or facilitates translesional syntheses catalyzed by mammalian DNA polymerases, mainly in an error-prone manner (19). Some of these polymerases such as pol η were able to catalyze significant amounts of lesion-bypass with miscoding opposite the adduct. This strongly suggests that 8-HM- ϵ C could be a promutagenic lesion in vivo, if such adducts are indeed formed in genomic DNA and not repaired.

8-HM- ϵ C is a structural analogue of the exocyclic adduct 3, N^4 -etheno C (ϵ C), in which the five-membered etheno ring has a $-\text{CH}_2\text{OH}$ group substituting a hydrogen (Figure 1). ϵ C is mainly formed from environmental compounds, such as vinyl chloride and ethyl carbamate, or through the process of lipid peroxidation (for review, see ref 20). Both in *E. coli* and mammalian systems, the ϵ C adduct in DNA has been shown to be miscoding in vitro and promutagenic in vivo (for review, see ref 21). In our recent study using mammalian DNA polymerases (19), ϵ C showed similar miscoding specificity to that of 8-HM- ϵ C.

In 1994, this laboratory first reported a DNA glycosylase activity in HeLa cell-free extracts which released ϵ C as a free base (22). During further purification of this activity, we found that it actually resides in a protein different from the glycosylase acting on 1, N^6 -ethenoadenine, namely, alkyl-purine-DNA- N -glycosylase (APNG) (23). This finding was further confirmed by the observation that the ϵ C-DNA glycosylase activity was unchanged in mouse tissues regardless of whether the APNG gene was knocked out (24). The protein excising ϵ C has now been identified as the mismatch-specific thymine-DNA glycosylase (TDG) in humans (25, 26) and mismatch uracil-DNA glycosylase (Mug) in *E. coli* (26), which is the homologue of the human TDG (27). One difference between these two enzymes is that Mug lacks the G \cdot T mismatch repair activity (27). The ϵ C activity was also observed in a recombinant G \cdot T mismatch DNA glycosylase from the thermophilic bacterium, *Methanobacterium thermoautotrophicum* THF (25), a functional homologue of TDG/Mug proteins (28). These enzymes were originally found to remove uracil or thymine from duplexes when paired with guanine (28–31). Both TDG and Mug are capable of excising ϵ C from oligomer duplexes efficiently (25, 26, 32), with Mug having greater efficiency toward the adduct (26).

In this work, we report that 8-HM- ϵ C is also a good substrate for the *E. coli* Mug protein. For comparative purposes, the previously identified substrate for the enzyme, ϵ C, was studied in parallel using in vitro assays. In addition, molecular modeling of these two closely related exocyclic derivatives in DNA duplexes was performed in order to obtain structural data which may aid in our understanding of enzymatic recognition of the two adducts.

EXPERIMENTAL PROCEDURES

Materials. Crotalus adamanteus venom phosphodiesterase 1, bacterial alkaline phosphatase, and [γ -³²P] ATP (specific activity 6000 Ci/mmol; 1 Ci = 37 GBq) were purchased from Amersham Pharmacia Biotech (Piscataway, NJ). T4 polynucleotide kinase was purchased from United States Biochemical (Cleveland, OH). Acetonitrile (HPLC grade) and glacial acetic acid were obtained from Fisher Scientific (Pittsburgh, PA). Triethylamine was purchased from Aldrich (St. Louis, MO). Sep-Pak C18 cartridges were purchased from Waters (Milford, MA). The unmodified phosphoramidites for ultramild deprotection, 4-isopropyl-phenoxy-acetyl (Pr-Pac)-dG, phenoxyacetyl(Pac)-dA, acetyl dC, and dT-CE phosphoramidite, and the dT ABI controlled pore glass (CPG) 500 Å column were purchased from Glen Research (Sterling, VA).

Repair Enzymes and Cell-Free Extracts. *E. coli* Mug protein, MutY protein, formamidopyrimidine-DNA glycosylase (Fpg protein), endonuclease III (EndoIII), and endonuclease VIII (EndoVIII) were from Trevigen (Gaithersburg, MD). One unit of Mug protein is defined as the amount of enzyme required to cleave 1 pmol of a ³²P-labeled oligomer containing an ϵ C in a duplex in 1 h at 37 °C. *E. coli* uracil-DNA glycosylase (Ung) was from Amersham Pharmacia Biotech. The major human AP endonuclease 1 (HAP1) was a gift from Dr. I. D. Hickson (University of Oxford, Oxford, U.K.). *E. coli* endonuclease IV (EndoIV) was a gift from Dr. D. M. Wilson (Lawrence Livermore National Laboratory, Livermore, CA). *E. coli* strain BW32 (AB1157), used to prepare a wild-type cell-free extract, was a gift from Dr. B. Weiss (University of Michigan, Ann Arbor, MI). The preparation of crude extract was carried out as previously described by Hang et al. (33).

8-HM- ϵ dC and Oligonucleotide Synthesis. The preparation of 8-HM- ϵ dC and its phosphoramidite was performed using a modification of the procedures described by Chenna et al. (18). The 25-mer oligodeoxynucleotide containing 3, N^4 -ethenocytosine (ϵ C) was synthesized as described previously (34). Unmodified oligodeoxynucleotides were synthesized and purified by Operon, Inc. (Alameda, CA).

The synthesis of the 25-mer 5'-CCGCTAGXGGGTAC-CGAGCTCGAAT-3' (X = 8-HM- ϵ C) was carried out on an Applied Biosystems 392 automated DNA synthesizer on a 1 μ mol scale, ABI column, using phosphoramidites which can be deprotected under ultramild conditions. The coupling time for the modified nucleoside was increased to 900 s to give an optimal coupling efficiency of 93%. The 5'-DMT-on 25-mer oligodeoxynucleotide was base-deprotected and cleaved from the resin, under strictly anhydrous conditions and in the absence of light, using 1,8-diazabicyclo[5.4.0]-undec-7-ene (DBU) dried over 3 Å molecular sieves with anhydrous methanol (100 μ L of DBU to 900 μ L of dry methanol). The deprotection was complete after 12 h stirring at room temperature.

Purification of the DMT-on oligomer was carried out on a Luna 5 μ phenyl hexyl 250 \times 4.60 mm analytical column (Phenomenex, Torrance, CA). The DMT-on oligomer was purified using a system in which the acetonitrile concentration was maintained at 15% for 10 min in the presence of triethylammonium acetate (TEAA) buffer, then increased linearly to 35% over the next 25 min. The flow rate was

¹ Abbreviations: 8-HM- ϵ C, 8-(hydroxymethyl)-3, N^4 -ethenocytosine; ϵ C, 3, N^4 -ethenocytosine; AP, apurinic/aprimidinic; Mug, mismatch uracil-DNA glycosylase; TDG, mismatch specific thymine-DNA glycosylase; Ung, uracil-DNA glycosylase; APNG, alkylpurine-DNA- N -glycosylase; PAGE, polyacrylamide gel electrophoresis; DMT, 4,4'-dimethoxytrityl chloride; TEAA, triethylammoniumacetate.

maintained at 1 mL/min. Using this system, the pure DMT-on oligomer had a retention time of 26.82 min.

The isolated DMT-on oligomer was treated with 80% acetic acid for 10 min, neutralized, evaporated to dryness, and purified again on the same column with the DMT-off. The resultant peak was collected using a second system where the acetonitrile was maintained at 5% in TEAA for 10 min and then increased linearly to 30% acetonitrile after 45 min. The retention time for the DMT-off oligodeoxynucleotide was 24.66 min. Analysis of this compound by electrospray mass spectrometry run on a VG Bio-Q Instruments mass spectrometer found the correct peak m/z : -7722.51 M^+ .

Further verification of the incorporation of the correct modified base was performed by enzyme digestion of the above 25-mer, according to the procedure defined in ref 35. The reaction mixture was then analyzed by HPLC (Hewlett-Packard 1100 detector) using a reversed-phase Luna 5 μ phenyl hexyl 250×4.60 mm analytical column. The result showed that by "spiking" the sample mixture with the modified monomer, an enlarged peak was observed superimposed over the modified monomer peak from the digested oligonucleotide at 19.39 min, confirming that the synthesized adduct was incorporated.

Band Shift and Glycosylase Assay. For testing protein binding and enzymatic activities, both modified and unmodified 25-mer oligonucleotides were 5'-end labeled with [γ - ^{32}P] ATP and annealed to a complementary oligonucleotide in a 1/1.5 ratio, as previously described (36). The standard binding reactions contained 1.5 nM 5'-end ^{32}P -labeled duplex in 20 mM Tris-HCl (pH 8.0), 1 mM EDTA, 1 mM EGTA, 0.1 mg/mL acetylated BSA, and varying concentrations of Mug protein in a total volume of 10 μL . After a 15 min incubation at room temperature ($\sim 20^\circ\text{C}$), reactions (5 μL) were resolved on 6% nondenaturing polyacrylamide gel electrophoresis (PAGE) for 1 h at 150 V using 1X TBE buffer.

To measure Mug excision activity, the same aforementioned reaction mixtures for the binding assay were incubated at 37°C with varying amounts of enzyme or for various lengths of time. The reactions were stopped by heating the samples at 95 – 100°C for 3 min and then placing on ice. Following this, a 5' AP endonuclease, HAP1 or EndoIV, was added to the reactions, which were incubated at 37°C for a further 20 min. This step was used to cleave the apyrimidinic (AP) site resulting from the excision of 8-HM- ϵC or ϵC by Mug. HAP1/EndoIV alone did not exhibit any detectable effect on these two substrates. In the reactions with AP lyase-containing glycosylases including Fpg protein, EndoIII, and EndoVIII, there was no addition of HAP1. One exception is MutY protein among the *E. coli* glycosylases tested. Whether this enzyme has an associated AP lyase activity has been the subject of controversy (e.g., refs 37–40). Therefore, HAP1 was added to the MutY reaction. When excision reactions were performed with cell-free extracts of *E. coli* AB1157, no AP endonuclease was added. For the competition assay, the same unlabeled 25-mer oligomer duplexes were preincubated with Mug protein in the mixture for 5 min on ice before ^{32}P -labeled 8-HM- ϵC -containing oligomer duplex was added. All of the above reactions were stopped by adding equal amounts of a F/E solution (90% formamide plus 50 mM EDTA) and then heated at 95 – 100°C for 3 min. Samples (5 μL /lane) were then run on 12% 8 M urea/PAGE,

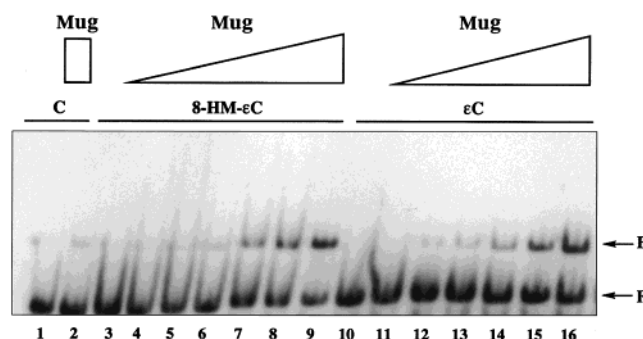


FIGURE 2: Binding of *E. coli* Mug protein to a 25-mer oligonucleotide duplex containing either 8-HM- ϵC or ϵC using a gel mobility shift assay. The reaction mixtures were incubated for 15 min at room temperature with increasing amounts of Mug protein. Lanes 4–9 and 11–16 contained 0.32, 0.8, 1.6, 3.2, 8.0, and 16 nM Mug protein. Lane 2 contained 16 nM Mug. Lanes 1, 3, and 10 contained buffer only. (B, bound; F, free DNA).

and subsequently, the gel was dried and autoradiographed. For band quantitation, the Bio-Rad FX Molecular Phosphor-Imager and Quantity One software (version 4.0.1) were used.

Molecular Modeling. The modified bases (8-HM- ϵC and ϵC) were constructed using the xLeap module of AMBER 5.0 (41). A set of parameters for the modified bases, including atom charges, bond and torsional angles, and bond stretching constants, have been developed based on ab initio quantum mechanical calculations, as described by Guliaev et al. (42). The modified bases were incorporated into 15-mer DNA duplexes truncated from the 25-mer sequence (see 8-HM- ϵC and oligonucleotide synthesis): 5'-CCGCT-AGXGGGTACC-3', where X = 8-HM- ϵC , ϵC , or C. The topology and coordinate files for all three duplexes were generated using the xLeap module of AMBER 5.0. Twenty-eight Na^+ ions were placed around phosphate groups to neutralize negative charges, and a rectangular water box was added which provided at least 10 Å of explicit water molecules around each of the DNA duplexes. The system was subjected to a series of equilibration runs to obtain the correct density and volume for the water present. Finally, 1 ns of the unrestrained molecular dynamics runs, using particle mesh Ewald (PME) to treat electrostatic interactions, were used to generate averaged structures (42). The average structures from the last 650 ps of MD were analyzed using CURVES 5.3 (43), and MD trajectories were analyzed using the carnal module of AMBER 5.0. The structures were also displayed and analyzed using Insight II (Insight II 98.0, Biosym/MSI, San Diego, CA). All calculations were performed on Silicon Graphics Origin 200 server (Silicon Graphics Inc., Mountain View, CA).

RESULTS

Recognition and Excision of 8-HM- ϵC by *E. coli* Mug Protein. We started our initial binding experiments on 8-HM- ϵC using *E. coli* mismatch uracil-DNA glycosylase (Mug), since this is the only enzyme found so far to remove ϵC in *E. coli* (26, 44). As shown in Figure 2, Mug protein showed a similar protein concentration-dependent binding activity toward a ^{32}P end labeled 8-HM- ϵC -containing oligomer duplex (lanes 4–9) to that toward ϵC -containing duplex (lanes 11–16). In this 25-mer, both 8-HM- ϵC and ϵC are in the same position (eighth nucleotide from 5' end), and the opposite base is G (see Experimental Procedures).

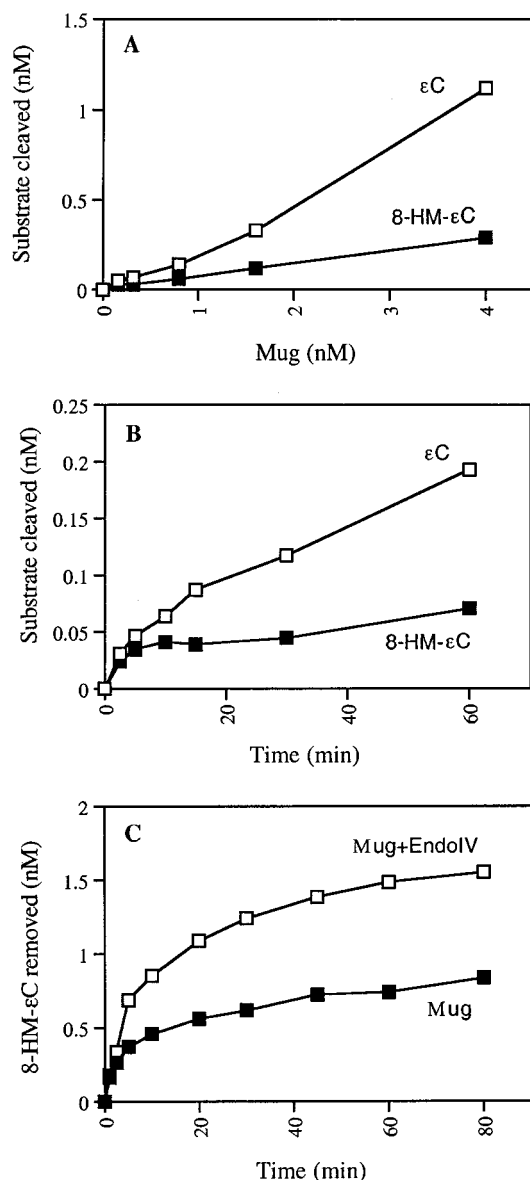


FIGURE 3: (A) Protein-dependent cleavage of 25-mer oligonucleotide containing either 8-HM- ϵ C or ϵ C by *E. coli* Mug protein. Increasing concentrations of Mug protein (0–4 nM) were incubated with 32 P end labeled oligomer substrates (1.5 nM) for 1 h at 37 °C. (For details, see Experimental Procedures.) (B) Time-dependent cleavage by *E. coli* Mug protein of 32 P end labeled 25-mer oligonucleotide containing either 8-HM- ϵ C or ϵ C. The oligomer duplexes (2 nM) were reacted with 0.5 nM Mug protein for varying times at 37 °C. (C) Stimulation of Mug activity toward 8-HM- ϵ C by *E. coli* EndoIV. Four nanomolar 32 P end labeled 25-mer duplex was incubated with 1 nM Mug protein with or without 4 nM EndoIV at 37 °C. One millimolar MgCl_2 was used for these reactions. Reactions were stopped at various time points by heating at 95–100 °C for 3 min. Ten nanomolar EndoIV was then added to all samples, including those with Mug plus EndoIV before heating. After 20 min at 37 °C, reactions were terminated by adding an F/E solution.

Mug protein was also shown to cleave the same 25-mer containing 8-HM- ϵ C in a protein concentration-dependent manner, but the extent of cleavage is lower than that of ϵ C cleavage (Figure 3A). In these experiments, the cleavage products from both substrates after a 5' AP endonuclease treatment were a 32 P-labeled 7-mer, resulting from hydrolysis 5' to the AP site at the eighth position. Alkaline treatment (NaOH plus heat) of Mug-treated samples also caused

incision of both 8-HM- ϵ C and ϵ C oligomer substrates. The cleavage pattern for the two adducts (data not shown) is the same, which is also in agreement with that described for the β -elimination reaction that cleaves an AP site from the 3' end (29, 45). Both experiments indicate that Mug acts on 8-HM- ϵ C as a DNA glycosylase.

The specificity of the 8-HM- ϵ C activity of Mug was further confirmed using a competition assay. As shown in Figure 4, when the 32 P end labeled 8-HM- ϵ C 25-mer duplex was incubated with an increasing molar excess (1x, 2.5x, 5x, and 10x) of the same but unlabeled competitor 25-mers, Mug activity toward 8-HM- ϵ C could be efficiently competed by the 8-HM- ϵ C- or ϵ C-containing duplex but not by the unmodified duplex. The cross-competition of 8-HM- ϵ C excision by the ϵ C oligomer confirmed that Mug is the enzyme acting on 8-HM- ϵ C. Note that, in Figure 4, the ϵ C-containing 25-mer duplex (lanes 7–10) appears to be a better competitor than the 8-HM- ϵ C-containing competitor (lanes 3–6), which can be correlated with the kinetic findings described here.

We further compared the kinetic features of Mug excision of 8-HM- ϵ C with those of ϵ C under the same assay conditions. Figure 3B shows the removal of 8-HM- ϵ C (2 nM) and ϵ C (2 nM) by Mug (0.5 nM) from the 25-mer duplexes as a function of time (0–60 min). Both activities had an early fast phase during the first 2.5 min, and then reached a slower rate afterward. The rate of 8-HM- ϵ C excision is slower than that of ϵ C excision in both phases, with \sim 2.5-fold difference after 15 min reaction (Figure 3B). Addition of *E. coli* EndoIV, a 5' AP endonuclease next to the Mug protein in *E. coli* BER pathway, stimulated the 8-HM- ϵ C excision by \sim 2-fold (Figure 3C). In both Figure 3B and C experiments, when a 4 molar excess of substrate duplex was used, the 8-HM- ϵ C or ϵ C excised by Mug in the absence of EndoIV was less than the molar concentrations of Mug in these reactions. However, in the addition of 4 nM EndoIV to the reaction in Figure 3C, which had 4 nM oligomer duplex and 1 nM Mug protein, more than 1 nM 8-HM- ϵ C was excised, indicating the occurrence of turnover with the substrate.

To explore whether any other glycosylase in *E. coli* can act on 8-HM- ϵ C, various available pure DNA glycosylases were tested. Except for Mug protein, no detectable activities were observed from other glycosylases tested (Figure 5). This test was carried out using high amounts of proteins compared to that required to cleave their normal substrates in order to detect weak activity. Figure 5 also shows that this newly identified 8-HM- ϵ C activity resides in a wild-type *E. coli* strain. The same pattern was also observed for ϵ C substrate specificity.

Effect of Opposite Bases and Strandness on Excision Efficiency. The adduct-containing 25-mers were 5'-end 32 P-labeled and annealed to a complementary strand with each of the four bases opposite 8-HM- ϵ C or ϵ C. Three *E. coli* DNA glycosylases were tested, including Mug, Ung, and MutY. Mug and MutY are known to recognize base mismatches. Table 1 summarizes the base pair specificity of the Mug protein, as determined by the initial velocities of the adduct excision. All modified base mismatches were recognized and the adduct excised by Mug. For both adducts, the variation in activities toward these mismatches was within a relatively small range (Table 1). Unlike Mug, neither Ung

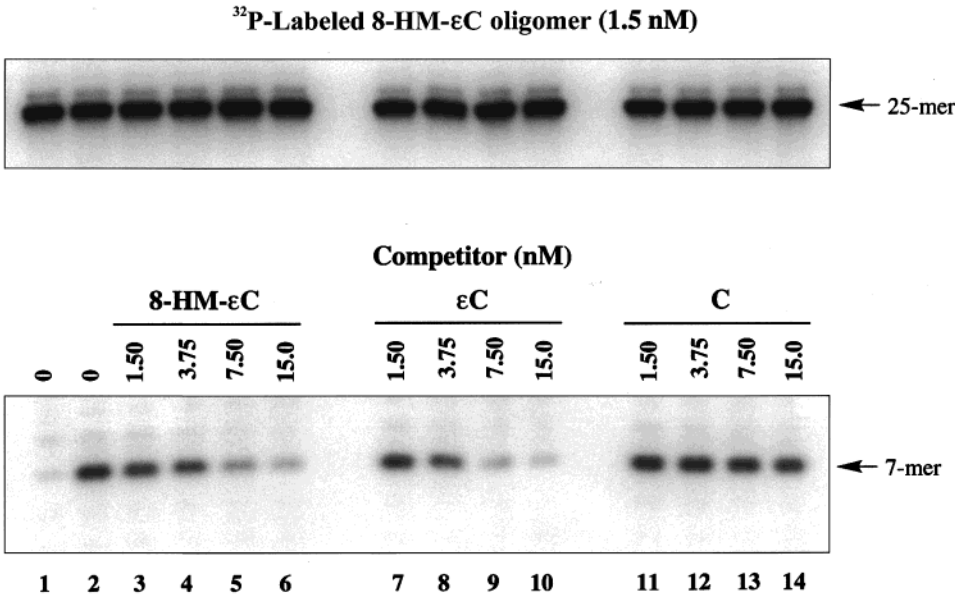


FIGURE 4: Mug excision of 8-HM-εC with and without competitors. The standard reaction mixtures were incubated for 10 min at 37 °C with increasing amounts of unlabeled 25-mer oligomer duplex containing either 8-HM-εC (lanes 3–6), εC (lanes 7–10), or unmodified C (lanes 11–14). Lane 1 contained buffer only. Lanes 2–14 contained 4 nM Mug protein.

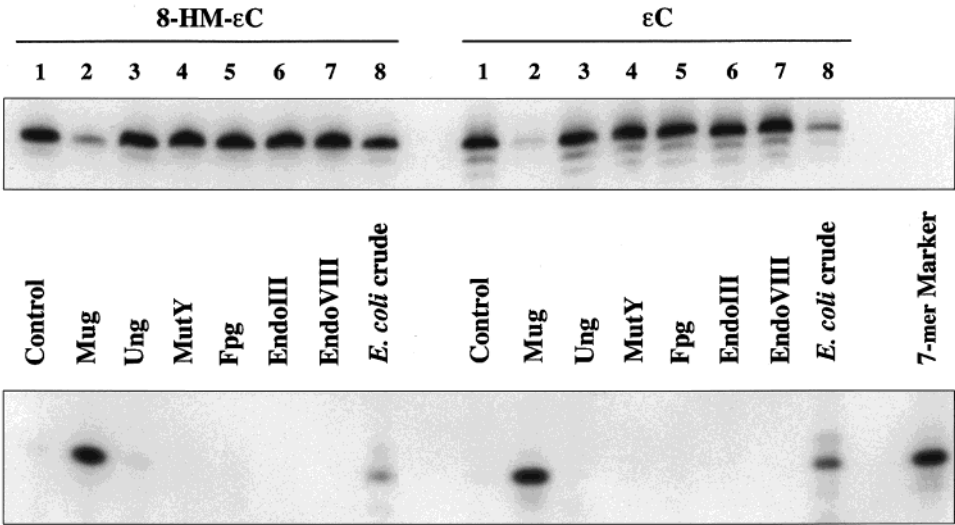


FIGURE 5: Screening of activities of *E. coli* DNA glycosylases toward oligomers containing either 8-HM-εC or εC. The substrate oligomer duplexes were incubated for 1 h at 37 °C with each individual purified DNA glycosylases (Mug: 0.01 U; Ung: 0.1 U; MutY: 0.1 U; Fpg protein: 0.1 U; EndoIII: 0.1 U; EndoVIII: 0.1 U). In addition, a cell-free extract prepared from a wild-type *E. coli* strain, BW 32, was also tested for activities against the modified bases (lanes 8). Lanes 1–4 in both panels also contained 5 ng of HAP1.

Table 1: Recognition of 8-HM-εC or εC Mispairs by Mug Protein ^a					
opposite base	8-HM-εC	εC	opposite base	8-HM-εC	εC
G	100 ^b	100	T	74	143
A	62	111	C	95	137

^a The standard reaction conditions were used for this experiment (see Experimental Procedures). The incubation time was 10 min at 37 °C. Data are an average of three independent experiments. No activity was detected using MutY or Ung for these mispairs. ^b Relative excision activity. The initial rates of excision activity measured on 8-HM-εC or εC were treated as 100%.

nor MutY showed any detectable activity toward any of the modified base mismatches listed in Table 1, showing a clear difference in their substrate specificity from Mug.

With the single-stranded oligomer containing either 8-HM-εC or εC, no detectable cleavage was observed when up to 16 nM Mug protein was used for 1 h at 37 °C (data not

shown). For these experiments, after Mug reaction with the adduct-containing oligonucleotide, a 2-fold molar excess of the complementary strand was added to the reaction which was then slowly cooled from 80 °C to room temperature. HAP1 (5 ng) was then added and incubated at 37 °C to cleave any AP site in duplex DNA, as this enzyme does not cleave AP site in single-stranded DNA (46).

Structural Features of 8-HM-εC and εC. The stability and the equilibrium state of the 1 ns MD simulations were evaluated by calculating the RMSD values of each 1 ps “snapshot” relative to the starting structures containing dC, 8-HM-εC, or εC. All three structures reached the conformational equilibrium in the range between 250 and 350 ps (data not shown). Therefore, the conformations generated between 350 and 1000 ps were used to monitor structural properties. All three structures remained in the B DNA conformational family, with the minor distortion around the

Table 2: Intrabase Pair Parameters for the C•G, ϵ C•G, and 8-HM- ϵ C•G Basepairs

basepair	shear (Å)	stretch (Å)	stagger (Å)	buckle (deg)	propeller (deg)	opening (deg)
C•G	0.15	0	−0.02	8.12	−0.11	−0.86
8-HM- ϵ C•G	1.38	1.0	−0.74	−5.99	−14.26	28.06
ϵ C•G	2.01	0.7	0.1	5.11	−14.04	33.47

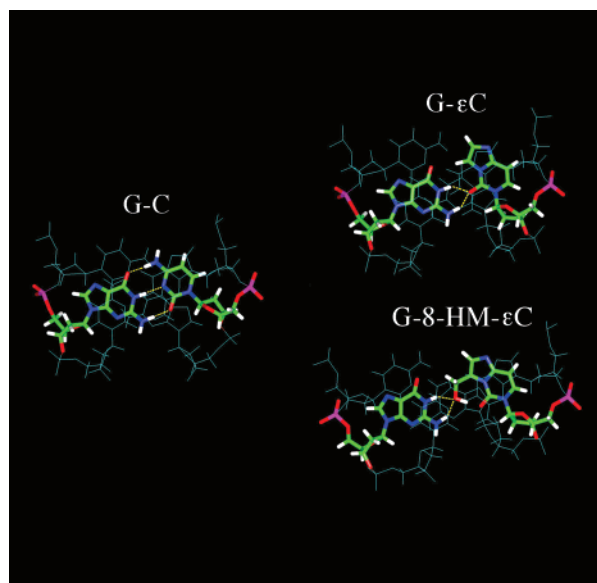


FIGURE 6: Top view of the central 3 basepair motif for the control (C•G) and two lesion-containing DNA duplexes (ϵ C•G and 8-HM- ϵ C•G, respectively) produced by molecular modeling. The central C•G, ϵ C•G, and 8-HM- ϵ C•G basepair are colored by atom type. The hydrogen bond patterns for each base-pair are shown in dotted yellow lines. Note the similar displacement of the adducts toward the major groove and similar sugar conformations in the ϵ C•G and 8-HM- ϵ C•G basepairs.

lesion site during the entire course of the simulation. The dC adducts were displaced into the major groove of the helix, while the opposite base remained stacked.

The main feature of the two lesion-containing basepairs is the similar alignment of the bases in the pair with the high value for the shear (SHR) parameter (Table 2). The SHR values for the other basepairs in all three DNA duplexes, including control, fall into the range -0.2 to 0.3 Å. The high SHR value correlated with the higher degree of opening of the lesion-containing basepair (Table 2). The alignment of the bases in the 8-HM- ϵ C•G and ϵ C•G pairs was stabilized by the observation of the single bifurcated hydrogen bond (yellow dotted lines in Figure 6). The stability of this bond was evaluated by calculating the percent of occupancy during the simulation. For the 8-HM- ϵ C•G pair, the hydrogen bond between 8-HM- ϵ C–O3 and G23–N2/N1 was 99.5% occupied with an average value of 3.98 ± 0.18 Å. In the case of ϵ C•G basepair, the hydrogen bond between ϵ C–O2 and G23–N2/N1 was 95% occupied with an average value of 3.01 ± 0.2 Å during the entire course of simulation. The observation of the highly sheared basepair for the ϵ C containing duplexes was reported previously using NMR (47–50). These authors showed that ϵ C opposite G, T, or C formed a sheared base-pair with one hydrogen bond between the bases.

Molecular dynamics calculations also revealed a similar sugar conformation for the two adducts. These adducts, when present in the DNA duplex, have a sugar pucker in the C3'-endo/C4'-exo region, while the rest of the residues, including

unmodified C, are in the C2'-endo/C3'-exo range (Figure 6). The C3'-endo/C4'-exo sugar conformation of the ϵ C adduct produced by modeling was in a good agreement with the previously reported solution structures of the ϵ C containing DNA duplexes, which showed the same conformational range for that sugar (48, 51).

DISCUSSION

Many chemicals can interact with cellular macromolecules and subsequently affect their functions. One of the most important factors in the initiation of carcinogenesis is the reaction of such chemicals with DNA bases to produce adducts or other types of damage (52). Glycidaldehyde, an animal carcinogen, has been shown to have the ability to modify bases both chemically (11–17) and in animal experiments (8, 13). In this work, we studied the enzymatic repair of 8-(hydroxymethyl)-3,*N*⁴-ethenocytosine (8-HM- ϵ C) as it is a potential glycidaldehyde-derived adduct in light of the previous work on the formation of hydroxymethyl etheno dA in vitro and in vivo (8, 11–16). In in vitro primer extension assays, 8-HM- ϵ C has been found to be a miscoding lesion when mammalian DNA polymerases were tested (19). In addition, the adduct itself is a structural analogue of 3,*N*⁴-ethenocytosine (ϵ C) (Figure 1), a well characterized exocyclic adduct (for review, see ref 53). It is of great interest for us to understand how the changes in adduct structure affect enzymatic recognition or catalytic efficiency (for review, see ref 54).

In this study, we first addressed the question whether the *E. coli* Mug protein, which excises ϵ C (26, 32, 44), also acts on the newly synthesized 8-HM- ϵ C. As shown in Figure 2, Mug is capable of binding to an 8-HM- ϵ C-containing oligomer duplex in a protein concentration-dependent manner, similar to the binding of ϵ C-duplex. Previous studies (32, 55–58) have revealed that both Mug and TDG bind strongly to their reaction product (i.e., an AP site-containing DNA when U•G- or T•G-containing substrates are used). In our binding experiments with the two exocyclic adducts, it is anticipated that the mechanism would be similar. Therefore, the binding of Mug to both adduct-containing oligomers in Figure 2 could be part of a process involving the initial recognition and also should reflect the generation of an AP site. Nevertheless, the latter point concerning the AP site binding should serve as indirect evidence that Mug possesses a DNA glycosylase activity toward both adducts.

Using an oligomer cleavage assay, Mug protein showed a DNA glycosylase activity toward 8-HM- ϵ C by forming an AP site at the adduct position. The rate of 8-HM- ϵ C excision is ~ 2.5 -fold lower than that of ϵ C excision (Figure 3B). These two activities showed a similar pattern in their rate change during the course of their reaction to that observed with a G•U-containing duplex (32) (i.e., the initial fast phase and a slow phase afterward). In the absence of *E. coli* EndoIV, Mug protein could not process more than its molar concentrations of the substrates (Figure 3B and C).

This is due to the known fact that Mug strongly binds to the reaction product AP site-containing DNA, resulting in the lack of Mug turnover (32). Addition of EndoIV to the Mug reaction, a BER enzyme which cleaves the AP site next to a glycosylase action, enhanced the Mug activity toward 8-HM- ϵ C by \sim 2-fold (Figure 3C). Moreover, the amount of 8-HM- ϵ C removed was more than the molar concentration (1 nM) of Mug (reached more than 1.5 nM after 60 min), clearly indicating the turnover for the substrate. The same mechanism has been studied previously for both Mug and TDG using G \cdot T(U)- and AP-site containing oligomer duplexes (32, 55, 57), which revealed that EndoIV/HAP1 displaces the tightly bound Mug/TDG from the AP site-containing oligonucleotide. It should be pointed out that these studies showed a higher efficiency of turnover than did our results after addition of an AP endonuclease, which could be due to the fact that Mug bound to an adduct-containing oligomer may be more "difficult" to be displaced by an AP endonuclease than that bound to a normal base mismatch such as G \cdot T(U).

The present finding that Mug also acts on 8-HM- ϵ C can be reasoned that this adduct closely resembles the structure of ϵ C. From the unrestrained molecular dynamics simulation, the alignment and hydrogen bond pattern of 8-HM- ϵ C-G pair is similar to that of ϵ C-G pair (Figure 6). The latter structure agrees in principle with that reported using NMR as described in the Results section, which would validate our computational approach. Both adducts formed a highly sheared pair with the opposite G and have the same puckered sugar conformation. Such a structural motif observed in ϵ C-containing duplexes was suggested to be important for its recognition by a specific DNA glycosylase (49).

The crystal structure of Mug complexed to an oligomer with a dU analogue-G mispair has been solved recently (59). From the same study using molecular modeling with ϵ C, it is understandable why ϵ C is a substrate for Mug. The authors showed that ϵ C can fit into the nonspecific pyrimidine pocket of the enzyme with its etheno moiety being comfortably accommodated in the hydrophobic space at the bottom of the pocket (59). In our case, the ability of Mug to recognize and remove 8-HM- ϵ C indicates that this ϵ C analogue should also fit into the active site. However, the lower Mug activity toward 8-HM- ϵ C suggests some degree of steric hindrance to the binding or catalytic activity as a result of the hydroxymethyl group on the etheno ring. Similar examples can be found from the literature (for review, see ref 54). One example is from the study by Pegg et al. (60) in which the authors reported that the rates of removal of *O*⁶-alkylguanines by *E. coli* *O*⁶-alkylguanine-DNA alkyltransferase were affected by the size of the residue. The larger the size of the residue, the slower the rate of the enzymatic activity toward the substrate (methyl > ethyl > hydroxyethyl).

For both 8-HM- ϵ C and ϵ C activities, like uracil excision, Mug protein requires double-stranded oligomer substrates. This enzyme differs from Ung protein in that Mug removes lesions only from duplex DNA (26, 32) (also designated as double-stranded uracil-DNA glycosylase (27)), while Ung has about 2-fold preference for excising U from a U-G pair in single-stranded DNA as compared to double-stranded DNA (61). When duplexes with modified base mismatches were used, Mug showed a minimal degree of preference toward any 8-HM- ϵ C or ϵ C mispairs (Table 1). With the ϵ C

adduct, previous NMR studies have shown that it cannot form strong hydrogen bonds with any opposite base (47–50). In contrast, none of these modified base mispairs is recognized by Ung or mutY. Ung has basically no sequence homology with Mug but shares structural similarity (62). Even though both enzymes recognize a U-G mismatch, Ung's tighter binding pocket seems to prevent the enzyme from accommodating ϵ C or 8-HM- ϵ C. The adenine-specific mismatch glycosylase, MutY, also seems not to act on any of the aforementioned modified base mispairs, indicating a strict requirement for its active site interaction.

Of all the *E. coli* glycosylases tested in this work, Mug is presently the only known *E. coli* enzyme excising 8-HM- ϵ C or ϵ C. The latter activity has been previously shown to be missing in cell-free extracts from a *mug* mutant (44), indicating that Mug may be the only glycosylase acting on ϵ C in *E. coli*. The ϵ C activity of Mug protein is known to be conserved during evolution because the homologous human TDG also excises ϵ C (25, 26). At present, there seems to be no way to determine whether these etheno and related derivatives are the original selective pressures under which Mug protein evolved or whether they are just fortuitously able to fit into the active site of an evolved protein. Considering the high miscoding potential of ϵ C as well as 8-HM- ϵ C, it is reasonable to assume that cells possess a specific repair enzyme toward these adducts.

REFERENCES

1. Knaap, G. A. C., Voogd, C. E., and Kramers, P. G. N. (1982) *Mutat. Res.* 101, 199–208.
2. Van Duuren, B. L., Langseth, L., Orris, L., Baden, M., and Kuschner, M. (1967) *J. Natl. Cancer Inst.* 39, 1213–1216.
3. Van Duuren, B. L. (1969) *Ann. N. Y. Acad. Sci.* 163, 633–651.
4. International Agency for Research on Cancer (1976) *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Man* (11), pp 175–181, IARC, Lyon, France.
5. International Agency for Research on Cancer (1987) *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Humans* (Suppl. 7), IARC, Lyon, France.
6. U.S. Department of Health and Human Services, National Toxicology Program (1998) *8th Report on Carcinogens* (1), III, pp 667–669, Washington, DC.
7. Bentley, P., Bieri, F., Kuster, H., Muakkassah-Kelly, S., Sagelsdorff, P., Staubli, W., and Waechter, F. (1989) *Carcinogenesis* 13, 321–327.
8. Steiner, S., Hönger, G., and Sagelsdorff, P. (1992) *Carcinogenesis* 13, 969–972.
9. International Agency for Research on Cancer (1985) *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Humans* (36), pp 181–188, IARC, Lyon, France.
10. International Agency for Research on Cancer (1989) *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Humans* (47), pp 237–261, IARC, Lyon, France.
11. Goldschmidt, B. M., Blazej, T. P., and Van Duuren, B. L. (1968) *Tetrahedron Lett.* 1583–1586.
12. Golding, B. T., Kennedy, G., and Watson, W. P. (1990) *Carcinogenesis* 11, 865–868.
13. Steiner, S., Crane, A. E., and Watson, W. P. (1992) *Carcinogenesis* 13, 119–124.
14. Van Duuren, B. L., and Loewengart, G. (1977) *J. Biol. Chem.* 252, 5370–5371.
15. Golding, B. T., Slaich, P. K., and Watson, P. (1986) *J. Chem. Soc., Chem. Commun.* 515–517.
16. Golding, B. T., Slaich, P. K., Kennedy, G., Bleasdale, C., and Watson, W. P. (1996) *Chem. Res. Toxicol.* 9, 147–157.
17. Kohwi, Y. (1989) *Carcinogenesis* 10, 2035–2042.

18. Chenna, A., Perry, A., and Singer, B. (2000) *Chem. Res. Toxicol.* 13, 208–213.
19. Singer, B., Medina, M., Zhang, Y., Wang, Z., Guliaev, A. B., and Hang, B. *Biochemistry*, in press.
20. Bartsch, H. (1999) in *Exocyclic DNA adducts in mutagenesis and carcinogenesis* (Singer, B., and Bartsch, H., Eds.), pp 1–16, IARC Scientific Publications (150), Lyon, France.
21. Barbin, A. (2000) *Mutat. Res.* 462 (2–3), 55–69.
22. Dosanjh, M. K., Chenna, A., Kim, E., Fraenkel-Conrat, H., Samson, L., and Singer, B. (1994) *Proc. Natl. Acad. Sci. U.S.A.* 91, 1024–1028.
23. Hang, B., Chenna, A., Rao, S., and Singer, B. (1996) *Carcinogenesis* 17, 155–157.
24. Hang, B., Singer, B., Margison, G. P., and Elder, R. H. (1997) *Proc. Natl. Acad. Sci. U.S.A.* 94, 12869–12874.
25. Hang, B., Medina, M., Fraenkel-Conrat, H., and Singer, B. (1998) *Proc. Natl. Acad. Sci. U.S.A.* 95, 13561–13566.
26. Saparbaev, M., and Laval, J. (1998) *Proc. Natl. Acad. Sci. U.S.A.* 95, 8508–8513.
27. Gallinari, P., and Jiricny, J. (1996) *Nature* 383, 735–738.
28. Horst, J. P., and Fritz, H. J. (1996) *EMBO J.* 15, 5459–5469.
29. Neddermann, P., and Jiricny, J. (1993) *J. Biol. Chem.* 268, 21218–21224.
30. Neddermann, P., and Jiricny, J. (1994) *Proc. Natl. Acad. Sci. U.S.A.* 91, 1642–1646.
31. Neddermann, P., Gallinari, P., Lettieri, T., Schmid, D., Truong, O., Hsuan, J. J., Wiebauer, K., and Jiricny, J. (1996) *J. Biol. Chem.* 271, 12767–12774.
32. Sung, J.-S., and Mosbaugh, D. W. (2000) *Biochemistry* 39, 10224–10235.
33. Hang, B., Chenna, A., Sági, J., and Singer, B. (1998) *Carcinogenesis* 19, 1339–1343.
34. Dosanjh, M. K., Chenna, A., Kim, E., Fraenkel-Conrat, H., Samson, L., and Singer, B. (1994) *Proc. Natl. Acad. Sci. U.S.A.* 91, 1024–1028.
35. ABI 392 model DNA synthesizer manual, Appendix 1, pp 1–6.
36. Rydberg, B., Qiu, Z.-H., Dosanjh, M. K., and Singer, B. (1992) *Cancer Res.* 52, 1377–1379.
37. Michaels, M. L., Cruz, C., Grollman, A. P., and Miller, J. H. (1992) *Proc. Natl. Acad. Sci. U.S.A.* 89, 7022–7025.
38. Manuel, R. C., and Lloyd, R. S. (1997) *Biochemistry* 36, 11140–11152.
39. Williams, S. D., and David, S. S. (1998) *Nucleic Acids Res.* 26, 5123–5133.
40. Williams, S. D., and David, S. S. (2000) *Biochemistry* 39, 10098–10109.
41. Case, D. A., Pearlman, D. A., Caldwell, J. W., Cheatham, T. E., Ross, W. S., Simmerling, C. L., Darden, T. A., Merz, K. M., Stanton, R. V., Cheng, A. L., et al., (1997) *AMBER 5*. University of California, San Francisco, CA.
42. Guliaev, A. B., Sági, J., and Singer, B. (2000) *Carcinogenesis* 21, 1727–1736.
43. Lavery, R., and Sklenar, H. (1996) *CURVES 5.1. Helical analysis of irregular nucleic acids*, Laboratory for Theoretical Biochemistry, CNRS, Paris, France.
44. Lutsenko, E., and Bhagwat, A. S. (1999) *J. Biol. Chem.* 274, 31034–31038.
45. Maxam, A. M., and Gilbert, W. (1977) *Proc. Natl. Acad. Sci. U.S.A.* 74, 560–564.
46. Sági, J., Hang, B., and Singer, B. (1999) *Chem. Res. Toxicol.* 12, 917–923.
47. Cullinan, D., Korobka, A., Grollman, A. P., Patel, D. J., Eisenberg, M., and de los Santos, C. (1996) *Biochemistry* 35, 13319–13327.
48. Cullinan, D., Johnson, F., Grollman, A. P., Eisenberg, M., and de los Santos, C. (1997) *Biochemistry* 36, 11933–11943.
49. Cullinan, D., Eisenberg, M., and de los Santos, C. (1999) in *Exocyclic DNA adducts in mutagenesis and carcinogenesis* (Singer, B., and Bartsch, H., Eds.), pp 179–189, IARC Scientific Publications (150), Lyon, France.
50. Korobka, A., Cullinan, D., Cosman, M., Grollman, A. P., Patel, D. J., Eisenberg, M., and de los Santos, C. (1996) *Biochemistry* 35, 13310–13318.
51. Cullinan, D., Johnson, F., and de los Santos, C. (2000) *J. Mol. Biol.* 296, 851–861.
52. Singer, B., and Grunberger, D. (1983) *Molecular Biology of Mutagens and Carcinogens*, pp 1–347, Plenum Press, New York.
53. Leonard, N. J. (1992) *Chemtracts: Biochem. Mol. Biol.* 3, 273–297.
54. Singer, B., and Hang, B. (1997) *Chem. Res. Toxicol.* 10, 713–732.
55. Waters, T. R., and Swann, P. F. (1998) *J. Biol. Chem.* 273, 20007–20014.
56. Schärer, O. D., Nash, H. M., Jiricny, J., Laval, J., and Verdine, G. L. (1998) *J. Biol. Chem.* 273, 8592–8597.
57. Waters, T. R., Gallinari, P., Jiricny, J., and Swann, P. F. (1999) *J. Biol. Chem.* 274, 67–74.
58. Hardeland, U., Bentele, M., Jiricny, J., and Schär, P. (2000) *J. Biol. Chem.* 275, 33449–33456.
59. Barrett, T. E., Schärer, O. D., Savva, R., Brown, T., Jiricny, J., Verdine, G. L., and Pearl, L. H. (1999) *EMBO J.* 18, 6599–6609.
60. Pegg, A. E., Scicchitano, D., and Dolan, M. E. (1984) *Cancer Res.* 44, 3806–3811.
61. Bennett, S. E., Sanderson, R. J., and Mosbaugh, D. W. (1995) *Biochemistry* 34, 6109–6119.
62. Barrett, T. E., Savva, R., Panayotou, G., Barlow, T., Brown, T., Jiricny, J., and Pearl, L. H. (1998) *Cell* 92, 117–129.

BI011542B